

Stereoselective Synthesis of the C1–C13 Fragment of (+)-Discodermolide Using Asymmetric Allyltitanations

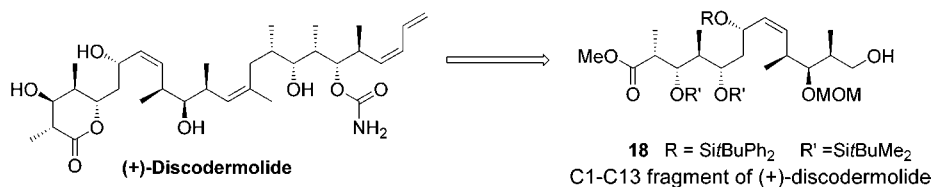
Samir BouzBouz* and Janine Cossy*

Laboratoire de Chimie Organique associé au CNRS, ESPCI, 10 rue Vauquelin
75231, Paris Cedex 05, France

janine.cossy@espci.fr; samir.bouzbouz@espci.fr

Received May 30, 2003

ABSTRACT



The synthesis of the C1–C13 fragment of (+)-discodermolide has been achieved. The configurations of the stereogenic centers have been controlled by enantioselective allyl- and crotyltitanations of aldehydes, and the Z configuration of the olefin at C8–C9 was controlled by a ring-closing metathesis.

Discodermolide was isolated by Gunasekera and co-workers at the Harbor Branch Oceanographic Institute in 1990 from the deep-water marine sponge *Discodermia dissoluta*.¹ This compound has a unique polyketide structure bearing 13 stereogenic centers, and the absolute configuration of these stereogenic centers was established by Schreiber et al. during their initial synthesis of both (+)- and (–)-discodermolide.

(+)-Discodermolide is both a potent immunosuppressive and anticancer agent as well as an antifungal agent.^{2,3} It inhibits T-cell proliferation with an IC₅₀ of 9 nM and graft-versus-host disease in transplanted mice.

Furthermore, startling cytotoxicity causing cell cycle arrest in the G2/M phase in a variety of human and marine cell lines.⁴ Discodermolide stabilizes microtubules and has been recognized as one of the most potent tubulin polymerizing

agents. Due to the potential therapeutic applications and the extreme scarcity of (+)-discodermolide (0.002% w/w) from frozen marine sponge, there has been considerable synthetic effort toward discodermolide, culminating in several total syntheses^{3,5} and numerous fragment syntheses.⁶ Herein, we report the results of our synthetic studies concerning an approach toward the C1–C13 subunit **18** which sets the stage for a convergent synthesis of (+)-discodermolide (Scheme 1).

According to our retrosynthetic analysis, the C14–C15 bond could be formed by using a sp²–sp³-type palladium(0)-mediated cross-coupling reaction between a vinyl

(1) (a) Gunasekera, S. P.; Gunasekera, M.; Longley, R. E. *J. Org. Chem.* **1990**, *55*, 4912; additions and corrections: *J. Org. Chem.* **1991**, *56*, 1346. (b) Gunasekera, S. P.; Pomponi, S. A.; Longley, R. E. U.S. Patent No. US5840750, Nov 24, 1998.

(2) (a) Longley, R. E.; Caddigan, D.; Harmody, D.; Gunasekera, M.; Gunasekera, S. P. *Transplantation* **1991**, *52*, 650. In vivo studies were also performed: (b) Longley, R. E.; Caddigan, D.; Harmody, D.; Gunasekera, M.; Gunasekera, S. P. *Transplantation* **1991**, *52*, 656. (c) Gunasekera, S. P.; Granik, S.; Longley, R. E. *J. Nat. Prod.* **1989**, *52*, 757.

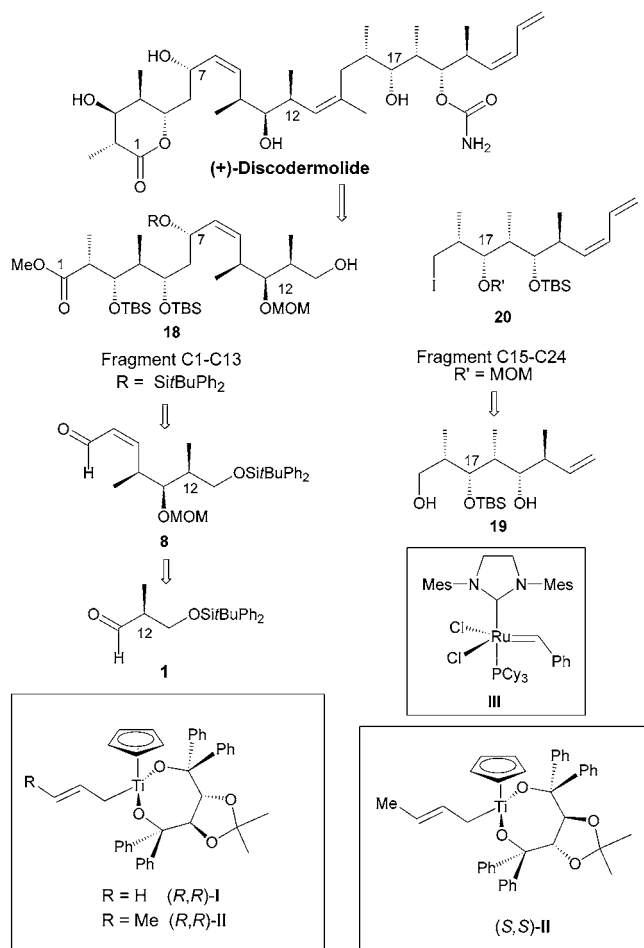
(3) (a) Nerenberg, J. B.; Hung, D. T.; Somers, P. K.; Schreiber, S. L. *J. Am. Chem. Soc.* **1993**, *115*, 12621. (b) Hung, D. T.; Nerenberg, J. B.; Schreiber, S. L. *J. Am. Chem. Soc.* **1996**, *118*, 11054.

(4) (a) Ter Haar, E.; Kowalski, R. J.; Hamel, E.; Lin, C. M.; Longley, R. E.; Gunasekera, S. P.; Rosenkranz, H. S.; Day, B. W. *Biochemistry* **1996**, *35*, 5, 243. (b) Hung, D. J.; Chen, J.; Schreiber, S. L. *Chem. Biol.* **1996**, *3*, 287.

(5) Total synthesis: (a) Smith, A. B., III; Qiu, Y.; Jones, D. R.; Kobayashi, K. *J. Am. Chem. Soc.* **1995**, *117*, 12011. (b) Smith, A. B., III; Kaufman, M. D.; Beauchamp, T. J.; LaMarche, M. J.; Arimoto, H. *Org. Lett.* **1999**, *1*, 1823. (c) Haried, S. S.; Yang, G.; Strawn, M. A.; Myles, D. C. *J. Org. Chem.* **1997**, *62*, 6098. (d) Marshall, J. A.; Lu, Z.-H.; Johns, B. A. *J. Org. Chem.* **1998**, *63*, 7885. (e) Paterson, I.; Florence, G. J.; Gerlach, K.; Scott, J. P. *Angew. Chem., Int. Ed.* **2000**, *39*, 2, 377. (f) Paterson, I.; Florence, G. J.; Gerlach, K.; Scott, J. P.; Sereinig, N. *J. Am. Chem. Soc.* **2001**, *123*, 9535. (g) Halstead, D. P. Ph.D. Thesis, Harvard University, Cambridge, MA, 1998. (h) Formal synthesis: Francavilla, C.; Chen, W.; Kinder, F. R., Jr. *Org. Lett.* **2003**, *5*, 1233.

(6) References for synthetic approaches to discodermolide, see: Arefolov, A.; Panek, J. S. *Org. Lett.* **2002**, *4*, 2397.

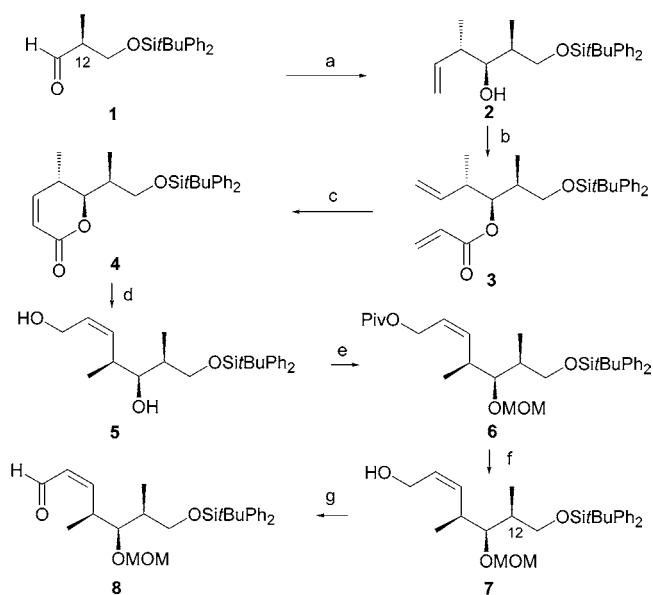
Scheme 1



iodide (C1–C14 fragment), which could be synthesized from compound **18** (C1–C13 fragment), and iodide **20** (C15–C24 fragment), which could be obtained from stereopentad **19**.⁷ Compound **18** could be synthesized from the unsaturated aldehyde **8**. We projected that the transformation of **8** to **18**, which implies the control of stereogenic centers C2, C4, C5, and C7, could be realized using the high face selective and stereoselective chiral allyltitanium complex (*R,R*)-**I** and the (*E*)-crotyltitanium complexes (*R,R*)-**II** and (*S,S*)-**II**.⁸ Aldehyde **8** will be synthesized from the commercially available (*S*)-2-methyl-3-silyloxypropanal **1** using the (*E*)-crotyltitanium complex (*R,R*)-**II** to set the C10 and C11 stereogenic centers. A ring-closing metathesis (RCM) reaction will control the *Z* double bond present at C8–C9 (Scheme 1).

The synthesis of the C1–C13 fragment of (+)-discodermolide began with the addition of the cyclopentadienyl-alkoxycrotyltitanium complex (*R,R*)-**II** to (*S*)-2-methyl-3-silyloxypropanal **1**. This addition afforded a 96/4 mixture of two separable diastereomers from which compound **2** was isolated in 91% yield. By analogy with previous results, the (*S*)-configuration can be attributed to the C11 and C10 stereogenic centers.⁹ To control the (*Z*) stereochemistry of

the double bond at C8–C9, compound **2** was transformed to lactone **4** in two steps. Compound **2** was first converted to the unsaturated ester **3** in 89% yield by treatment with acryloyl chloride in the presence of *i*-Pr₂NEt (CH₂Cl₂, –78 °C). Subsequent ring-closing metathesis using Grubbs' catalyst **III**¹⁰ in refluxing CH₂Cl₂ afforded lactone **4** (85% yield). This lactone was transformed to the allylic alcohol **5** in 72% yield by reduction with Dibal-H at –78 °C followed by reduction of the subsequent lactol by NaBH₄ in MeOH at 0 °C. It is worth noting that the ring-closing metathesis of an unsaturated ester followed by reduction of the lactone resulted in the exclusive formation of the (*Z*)-allylic alcohol **5**. Protection of this allylic alcohol as a pivaloyl ester (PivCl, pyridine, 25 °C) followed by protection of the secondary alcohol as a MOM ether (MOMCl, *i*-Pr₂NEt, CH₂Cl₂, 25 °C) led to compound **6** in 81% yield. To transform **6** to aldehyde **8**, compound **6** was first reduced with Dibal-H (CH₂Cl₂, –78 °C) to the corresponding alcohol **7** and then transformed to aldehyde **8** with a Swern oxidation (90% yield) (Scheme 2).

Scheme 2^a

^a Key: (a) (*R,R*)-**II**, ether, –78 °C, 91%; (b) acryloyl chloride, *i*-Pr₂NEt, CH₂Cl₂, –78 °C, 89%; (c) Grubbs' catalyst **III**, CH₂Cl₂, reflux, 85%; (d) (i) Dibal-H, CH₂Cl₂, –78 °C, (ii) NaBH₄, MeOH, 0 °C, 72%; (e) (i) PivCl, pyridine, 25 °C, (ii) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, 25 °C, 81%; (f) Dibal-H, CH₂Cl₂, –78 °C, 93%; (g) (COCl)₂, DMSO, –78 °C, Et₃N, CH₂Cl₂, 90%.

The elaboration of a suitable precursor to the C1–C13 fragment of (+)-discodermolide from **8** required the installation of the C7 stereogenic center with the (*S*)-configuration and, thus, the use of the allyltitanium complex (*R,R*)-**I** in ether at –78 °C. The homoallylic alcohol **9** was obtained in

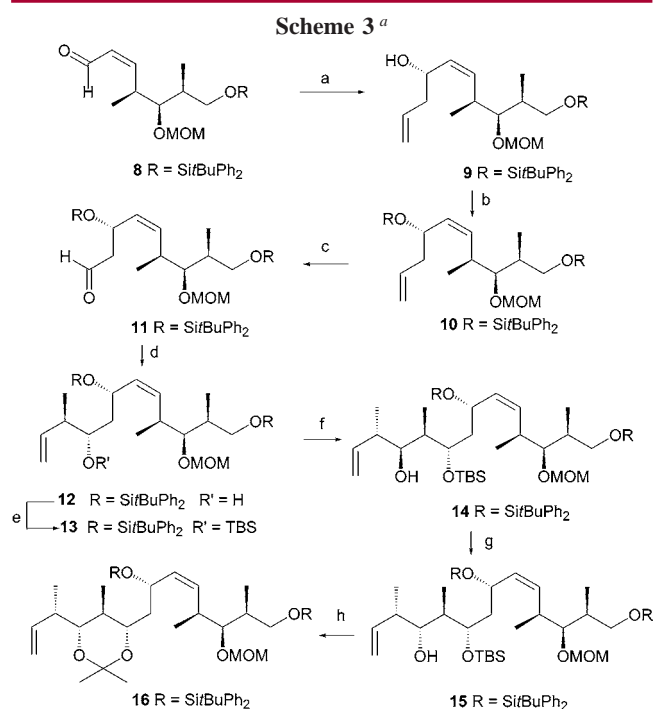
(7) BouzBouz, S.; Cossy, J. *Org. Lett.* **2001**, *3*, 3995.

(8) Hafner, A.; Duthaler, R. O.; Marti, R.; Rihs, J.; Rothe-Streit, P.; Schwarzenbach, F. *J. Am. Chem. Soc.* **1992**, *114*, 2321.

(9) The structure of **2** was determined by comparison with the analytical data reported in the literature see: Roush, W. R.; Palkowitz, A. D.; Ando, K. *J. Am. Chem. Soc.* **1990**, *112*, 6348.

(10) Grubbs, R. H.; Scholl, M.; Ding, S.; Lee, C. W. *Org. Lett.* **1999**, *1*, 953.

high diastereoselectivity ($dr = 96/4$) and in 90% yield by treatment of **8** with the (*R,R*)-**I** complex. The choice of a hindered protecting group for the hydroxy group is of significant importance for the progress of the synthesis because it will protect the disubstituted double bond at C8–C9 during the oxidation of the terminal olefin.¹¹ Thus, the homoallylic alcohol **9** was subjected to TBDPSCl (imidazole, CH_2Cl_2) and transformed to the silyl ether **10** in 93% yield. As anticipated, the oxidation of **10** with $\text{OsO}_4/\text{NMO}/\text{NaIO}_4$ (acetone/ H_2O) was selective and gave exclusively aldehyde **11** in high yield (89%). Therefore, the presence of the aldehyde was appropriate for the introduction of the C4 and C5 stereogenic centers which were controlled by using the (*S,S*)-**II** complex at -78°C . After treatment of **11** with the (*S,S*)-**II** complex, compound **12** was obtained in high diastereoselectivity ($dr = 95/5$) and in good yield (91%). Protection of the hydroxy group at C5 with TBSOTf (2,6-lutidine, -78°C) afforded compound **13** in 85% yield. The last two stereogenic centers at C2 and C3 were introduced in a three-step sequence from compound **13** (Scheme 3). The



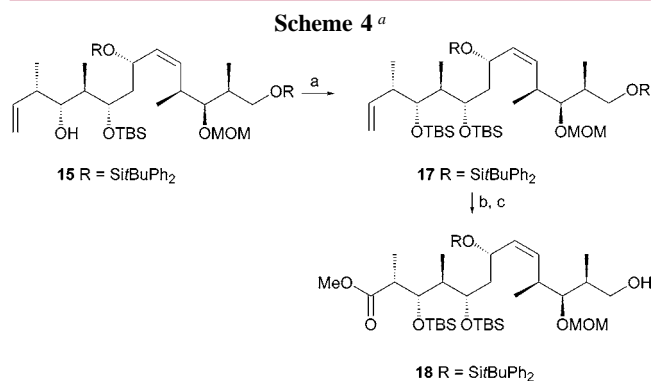
^a Key: (a) (*R,R*)-**I**, ether, -78°C , 90%; (b) ClSi-t-BuPh_2 , imidazole, CH_2Cl_2 , 93%; (c) $\text{OsO}_4/\text{NMO}/\text{NaIO}_4$, acetone/ H_2O , 89%; (d) (*S,S*)-**II**, ether, -78°C , 91%; (e) TBSOTf, 2,6-lutidine, -78°C , 85%; (f) (i) $\text{OsO}_4/\text{NMO}/\text{NaIO}_4$, acetone/ H_2O , (ii) (*R,R*)-**II**, ether, -78°C , 78%; (g) (i) $(\text{COCl})_2$, DMSO, -78°C , Et_3N , CH_2Cl_2 , (ii) Dibal-H, toluene, -80°C , 81%; (h) (i) TBAF, THF, 25°C , 24 h, (ii) ClSi-t-BuPh_2 , imidazole, CH_2Cl_2 , (iii) DMP, acetone, CSA, 25°C , 60%.

terminal double bond present in **13** was selectively oxidized to the corresponding aldehyde ($\text{OsO}_4/\text{NMO}/\text{NaIO}_4$), which was then treated directly with the (*R,R*)-**II** complex (ether, -78°C) to produce the homoallylic alcohol **14** in an overall

(11) Andrus, M. B.; Lepore, S. D.; Sclafani, J. A. *Tetrahedron Lett.* **1997**, 38, 4043.

yield of 78%, with the *anti* relative stereochemistry between the groups at C2 and C3. Inversion of the hydroxy group at C3 was achieved by using a Swern oxidation followed by a diastereoselective reduction of the obtained ketone by Dibal-H in toluene at -80°C .¹² This two-step process led to compound **15** in an overall yield of 81% (*syn/anti* > 25/1) (Scheme 3).

To establish the relative stereochemistry at C3, C4, and C5, **15** was transformed to acetonide **16** in three steps. After removal of the three silyl protecting groups, the allylic alcohol at C7 was protected selectively with TBDPSCl (imidazole, CH_2Cl_2), and the obtained product was treated with dimethoxypropane in acetone in the presence of CSA to produce acetonide **16**¹³ in 60% overall yield. The relative *syn* configuration of the hydroxy groups at C3 and C5 was confirmed by the analysis of the ^{13}C NMR ($\delta = 19.5, 30.0$ for Me_2C) (Scheme 3). The transformation of **15** to ester **18**, which corresponds to the C1–C13 fragment of discodermolide, was achieved in five steps. The protection of the hydroxy group at C8 as a TBS ether led to **17** in 84% yield. After selective cleavage of the terminal double bond ($\text{OsO}_4/\text{NMO}/\text{NaIO}_4$), the aldehyde was oxidized to the carboxylic acid upon treatment with buffered NaClO_2 . The methyl ester was produced by esterification of the acid (TMSCHN_2 , MeOH, benzene, 65% yield) and selectively deprotected to give alcohol **18** in 82% yield upon treatment with NH_4F in refluxing MeOH (Scheme 4).



^a Key: (a) TBSOTf, 2,6-lutidine, -20°C , 84%; (b) (i) OsO_4 , NMO, NaIO_4 , acetone/ H_2O , 25°C , (ii) NaClO_2 , NaH_2PO_4 , H_2O , *t*-BuOH/ H_2O , (iii) TMSCHN_2 , MeOH, benzene, 65%; (c) NH_4F , MeOH, 65°C , 82%.

In conclusion, the synthesis of the fully elaborated C1–C13 fragment of (+)-discodermolide, compound **8**, was completed by using high face selective chiral allyl- and crotyltitanations. Progress toward the total synthesis of (+)-discodermolide continues and will be reported in due course.

Supporting Information Available: ^1H NMR and ^{13}C NMR spectra for compounds **5**, **10**, **16**, and **18**. This material is available free of charge via the Internet at <http://pubs.acs.org>. OL034958L

(12) Boger, D. L.; Curran, T. T. *J. Org. Chem.* **1992**, 57, 2235.
(13) Rychnovsky, S. D.; Rogers, B. N.; Richardson, R. I. *Acc. Chem. Res.* **1998**, 31, 9.